

(19) World Intellectual Property Organization
International Bureau



(43) International Publication Date
20 December 2001 (20.12.2001)

PCT

(10) International Publication Number
WO 01/97314 A1

(51) International Patent Classification⁷: **H01M 8/24**, 8/04

(21) International Application Number: PCT/US01/40990

(22) International Filing Date: 13 June 2001 (13.06.2001)

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:
60/211,444 13 June 2000 (13.06.2000) US

(63) Related by continuation (CON) or continuation-in-part (CIP) to earlier application:
US 60/211,444 (CON)
Filed on 13 June 2000 (13.06.2000)

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(81) Designated States (*national*): CA, US.

(84) Designated States (*regional*): European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR).

Published:

— *with international search report*

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(54) Title: **REDUCED SIZE FUEL CELL FOR PORTABLE APPLICATIONS**

(57) Abstract: A flat pack type fuel cell includes a plurality of membrane electrode assemblies. Each membrane electrode assembly is formed of an anode, and electrolyte, and a cathode with appropriate catalysts thereon. The anode is directly into contact with fuel via a wicking element. The fuel reservoir may extend along the same axis as the membrane electrode assemblies, so that fuel can be applied to each of the anodes. Each of the fuel cell elements is interconnected together to provide the voltage outputs in series.



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REDUCED SIZE FUEL CELL FOR PORTABLE APPLICATIONS

The present application claims priority from provisional application number 60/211,444, filed June 13, 2000.

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The invention described herein was made in the performance of work under a NASA contract, and is subject to the provisions of Public Law 96-517(35 USC 202) in which the contractor has elected to retain title.

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Background

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Many applications exist for small power sources. For example, such devices have been used to power portable computers, cell phones, and the like. Conventional type sources may include rechargeable batteries. The current state-of-the-art is lithium ion batteries. These batteries have an energy content of approximately 150 watt-hours/kg. Once the energy content is exhausted, the user must recharge the battery. This means that users must either wait while the battery is recharged, or carry additional batteries, recharging equipment, and the like. The recharging can also only be carried out when the user is in proximity to a power outlet.

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Fuel cells are known in the art as devices which produce electricity when provided with fuel. However, conventional fuel cells include bipolar plate stacks, pumps, blowers, and other devices which may add considerable complexity to the final device.

Summary

The present application teaches a fuel cell apparatus which has a structure that is highly suitable for miniaturization. In one aspect, the apparatus has a structure that brings fuel into contact with specified parts of the fuel cell. A wicking structure may be used to bring the fuel into contact with the fuel cell element.

Brief description of the drawings

These and other aspects will now be described in detail with reference to the accompanying drawings, wherein:

Figure 1A shows a front on view of a flat pack fuel cell of an embodiment;

figure 1B shows an cross sectional edge view of a first embodiment of the fuel cell system along the line 1b-1b in figure 1a;

figure 2 shows a top view of the first embodiment;

Figure 3 shows a top view of an alternative embodiment;

Figure 4 shows an embodiment with multiple flat pack elements electrically connected together.

Detailed description

The figure 1A and 1B embodiment show an edge view of the miniaturized, flatpack type fuel cell. Electrical energy is generated by oxidation of organic fuel and reduction of oxygen in air to water. Any of a

number of different kinds of fuel cells may be used. One type is the fuel cell described in U.S. patent number 5,599,638.

The flatpack fuel cell may include a plurality of cells interconnected both in series and in parallel. Figure 1A shows a front view of the fuel cell assembly, with plural membrane electrode assemblies 97, 98, and 99. Figure 1B shows a cross section along the line 1b-1b, which shows the MEAs 97, 98, 99.

Each of the membrane electrode assemblies such as 97 includes an anode 105 and a cathode 110. The anode 105 and cathode 110 are separated by a polymer electrolyte membrane 115. A polymer electrolyte membrane may be, for example, of the type described in US Patent numbers 6,150,047 and 6,136,463. In this embodiment, all of the cells such as 97, 98, 99 are arranged along an axis 126, in a single contiguous plane.

As shown in figure 1b, the fuel cell elements are effectively connected in series. The cathode 103 of fuel cell element 97 is connected via the interconnects 135 to the anode 104 of the next fuel cell element 98. Similarly, each cathode is connected in series to the next anode. This provides the fuel cell elements being in series, providing outputs at the final electrodes 127, 128, that correspond to a series combination of all of the voltages.

The anode is in contact with a wicking structure 120 which also extends along that axis, substantially parallel to the anodes. The wicking structure 120 itself is in contact with the fuel source 125, which may also include a refillable fuel reservoir 130.

The wicking structure can be made of any absorbent material, such as

an absorbent pad, or any other absorbent material that is chemically stable in contact with the fuel and also electrochemically stable in contact with the anode. In fact, any structure that can provide the liquid fuel via capillary action can be used for this purpose. The fuel source and fuel reservoir may
5 hold the fuel that drives the electrochemical reaction. The fuel may be absorbed by the wicking structure, and provided to the anode.

Wicking structure 120 absorbs the fuel, and provides the fuel to the anodes. The wicking structure may deliver the fuel in regulated amounts to the surfaces of the anodes. A plurality of cathodes 110 are each in contact
10 with air, for the oxygen used in the electrochemical reaction.

The cells create a voltage by the electrochemical reaction. This voltage is produced between the top current collector 127 and the bottom current collector 128.

A top view of the cell is shown in figure 2. This also shows the
15 electrode/current collector 127, and the corresponding electrodes of the other cells 97, 98,99.

Conventional fuel cells have often used "bipolar plates" which may add weight, volume, complexity and cost to the fuel cell. These bipolar plates are not used in the flatpack cell. Instead, the cells are connected using interconnects
20 135, 136, that connect to and/or extend through part of the membrane. These interconnects may be made from corrosion resistant conductive materials. Example materials which can form a through-membrane connectors include graphite, platinum, gold, and appropriate stable polymer binders such as PVDF or Nafion.

A second embodiment, formed in figure 3, uses an edge connector instead of the through-membrane connectors. The edge and connector configuration is formed of a thin strip of conductive material. An insulator slot 300 is formed, and the interconnects 305,310 are formed in tabs in the
5 insulator slot. These interconnects are connected between the electrodes 315. The interconnect tabs may be formed of gold or graphite, for example.

The membrane electrode assemblies may be formed in the conventional way. The anodes in figure 1 may first be fabricated by applying catalyst layers and backing structures. The membrane is applied to the
10 anode, and the catalyst layer and cathode are fitted together. The catalyst may be applied, for example, using catalyst inks or catalyst-pre-coated membranes, or may be applied using a sputter deposition process. Gas diffusion backing layers may also be bonded to the membrane using a hot pressing process. An example is described in US Patent no 5,599,638; and
15 6,171,721.

An alternative, non-bonded backing layers can be used to form the membrane electrode assemblies. Other alternatives may include preparation of such assemblies by reactive sputter deposition of metal catalyst layers, spray deposition, chemical vapor deposition, electrodeposition, ion
20 impregnation, in situ catalyst deposition from organic metal precursor and combustion chemical vapor deposition.

A platinum-ruthenium catalyst may be used at the anode. The cathode may use a pure platinum catalyst, for example. However, other catalysts may be used which involve binary and ternary compositions of Pt, Ru, Ti, Zr, Ir and

Os, especially on the anode.

The anode structure may be made hydrophilic, so that liquid organic fuel may be absorbed through the anode. The wicking structure 120 brings the fuel into contact with the anode 105, allowing the liquid organic fuel to
5 access the catalyst layer, and to allow carbon dioxide product to readily leave the surface. The cathode may be rendered hydrophobic, in order to prevent water from saturating the electrode. This also provides air more ready access to the catalyst layer.

Two of the basic building blocks shown in figures 1-3 may be combined
10 in series or in parallel to increase the voltages. For example, a three cell flatpack may be capable of producing a terminal voltage in the range of 1 – 1.2 V depending on the load that is placed on the voltage. However, multiple ones of these arrays may be used to obtain higher voltages.

Figure 4 shows two sets of flat pack fuel cells 405,410 arranged with a
15 common fuel feed 400, e.g., a methanol feed, arranged between the two fuel cells. The two flatpacks each have respective outputs. The fuel cell element 405 has output terminals 401,402. Similarly, the fuel cell element 410 includes the outputs 411, 412. These output terminals may be connected
may be electrically connected in series to increase the voltage output. For
20 example, terminal 402 could be connected to terminal 411, with outputs being obtained between terminals 401 and 12. Alternatively, the fuel cells could be connected in parallel to increase the current handling capability.

The device may be made and tested in an enclosure with an internal absorbent pad on the anode side that retains the methanol solution. The fuel

is delivered to the anode via capillary action. The other end of the housing has multiple air openings allowing air access. However, other housings may be similarly used.

Although only a few embodiments have been disclosed in detail above,
5 other embodiments are possible. All such modifications are intended to be encompassed within the following claims, in which:

What is claimed is:

1. A fuel cell, comprising:

a plurality of fuel cell elements, including electrical connection parts that include an anode and a cathode, said plurality of fuel cell elements
5 extending along an extended axis;

a fuel reservoir, extending along said extended axis; and

a fuel delivery element, extending between said fuel reservoir and said anodes of all of said fuel cell elements, and supplying fuel to all of said anodes extending along said extended axis.

10

2. A fuel cell as in claim 1, wherein said fuel delivery element includes a wicking structure that provides fuel from said fuel reservoir to said anodes via capillary action.

15

3. A fuel cell as in claim 2, wherein said fuel reservoir extends along an entirety of said extended axis, from one end of said one electrical connection parts to the other end of said one electrical connection parts.

20

4. A fuel cell as in claim 3, wherein said anode is rendered hydrophilic, to allow liquid fuel to be more readily absorbed thereby.

5. A fuel cell as in claim 4, wherein said cathodes, are located on an opposite side of said plurality of fuel cell elements from said anode, and said cathodes are rendered hydrophobic.

6. A fuel cell as in claim 5, further comprising a housing, said housing including air holes in a vicinity of said cathode.

7. A fuel cell as in claim 1, further comprising a plurality of
5 interconnections between said fuel cell elements, said plurality of interconnections including electrical interconnections.

8. A fuel cell as in claim 7, wherein said plurality of interconnections are formed within said fuel cell, and are formed of a corrosion resistant and
10 electrically conductive material.

9. A fuel cell as in claim 7, wherein each of said fuel cell elements includes an anode material, a cathode material, a catalyst, and a membrane electrolyte material formed between said anode material and said cathode
15 material, and formed together to form a membrane electrode assembly.

10. A fuel cell as in claim 9, wherein said plurality of interconnections are formed through said membrane electrolyte material, between a membrane electrolyte material of one of said fuel cell elements, and a membrane
20 electrolyte material of another of said fuel cell elements.

11. A fuel cell as in claim 10, wherein said interconnections are formed from a corrosion resistant electrically conductive material.

12. A fuel cell as in claim 7, wherein said interconnections include a connector plate connecting between portions of said electrodes.

13. A fuel cell as in claim 1, further comprising a second plurality of fuel cell elements, formed along a second extended axis parallel from but spaced from said extended axis, and also in connection with said fuel reservoir.

14. A fuel cell as in claim 13, wherein said fuel delivery element includes a first wicking structure which extends between said fuel reservoir and anodes of said plurality of fuel cell elements, and a second wicking structure extending between second plurality of fuel cell elements and anodes of the second plurality of fuel cell elements, wherein said first and second wicking structures are each in contact with the single fuel reservoir.

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15. A fuel cell as in claim 7, wherein said interconnections extend between a cathode of a first fuel cell and an anode of a second fuel cell, effectively putting said first fuel cell and said second fuel cell electrically in series.

20

16. A method, comprising:

providing a plurality of fuel cell elements along an extended axis, each of said fuel cell elements including an anode, an electrolyte membrane, and a cathode;

providing fuel from a fuel reservoir to each of said cathodes of each of said fuel cell elements directly, via capillary action.

17. A method as in claim 16, further comprising electrically connecting
5 said plurality of fuel cell elements electrically in series with one another.

18. A method as in claim 17, wherein said electrically connecting comprises providing a corrosion resistant electrical interconnected between said fuel cell elements.

10

19. A method as in claim 17, further comprising a second plurality of fuel cell elements, extending along a second extended axis, and

providing fuel from said fuel reservoir to the other the second plurality of fuel cell elements, using capillary action from the single fuel in the fuel
15 reservoir.

20. A method as in claim 19, wherein said second plurality of fuel cell elements are in parallel with said plurality of fuel cell elements.

21. A method as in claim 17, wherein said electrically connected comprises providing an interconnected tab between electrodes.

20

22. A method as in claim 16, further comprising rendering the anode hydrophilic, to facilitate fuel of sorption into the anode.

23. A method as in claim 16, further comprising rendering the cathode hydrophobic, to prevent fuel from absorbing into the anode.

5 24. A method as in claim 22, further comprising rendering the cathode hydrophobic, to prevent fuel from absorbing into the cathode.

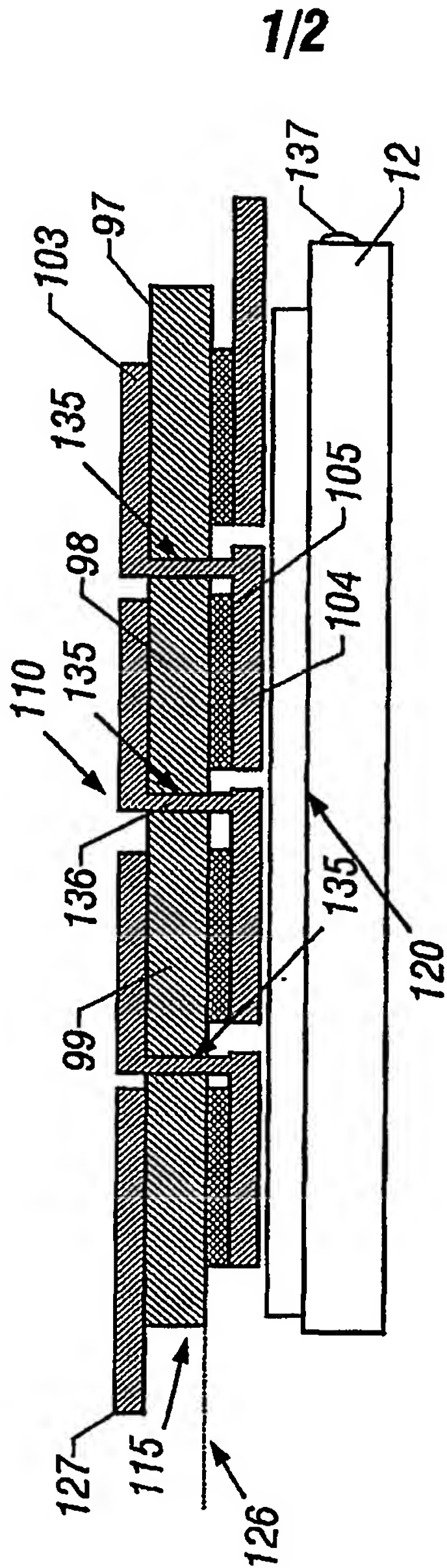
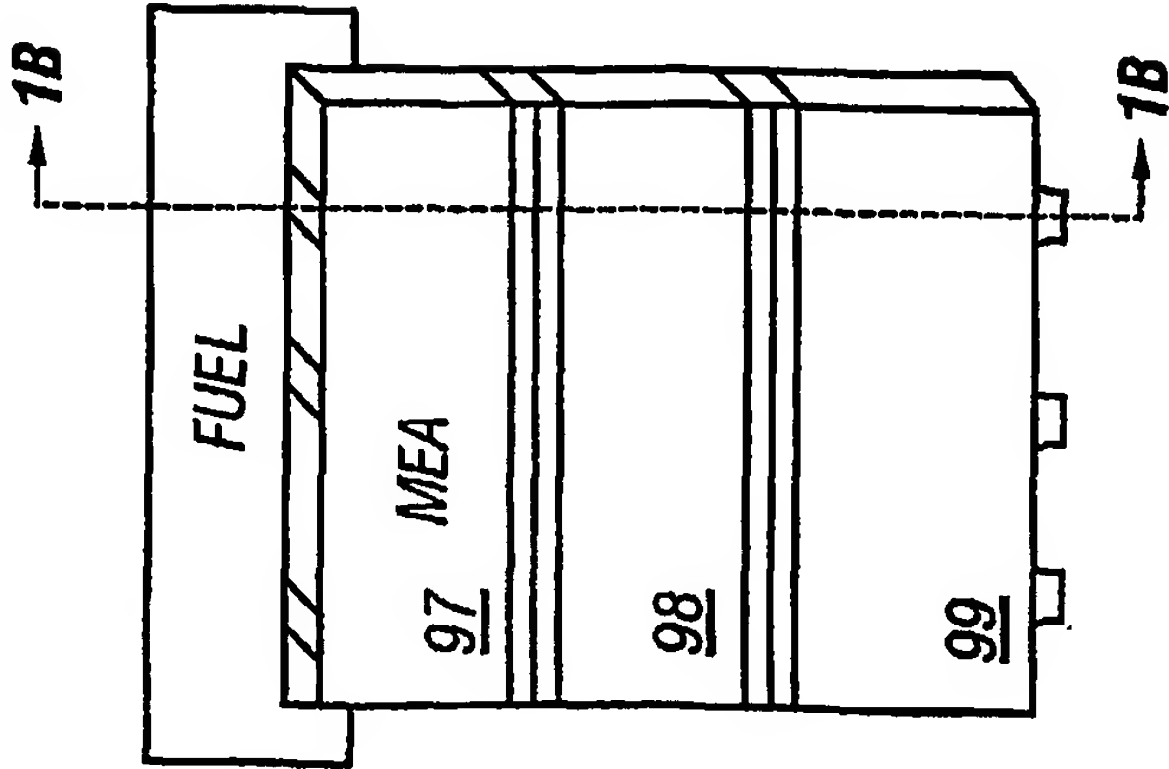


FIG. 1A

FIG. 1B

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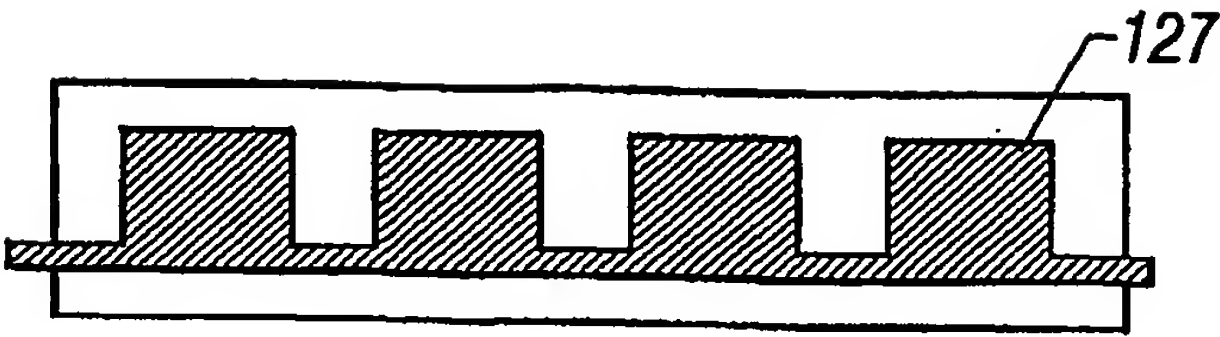


FIG. 2

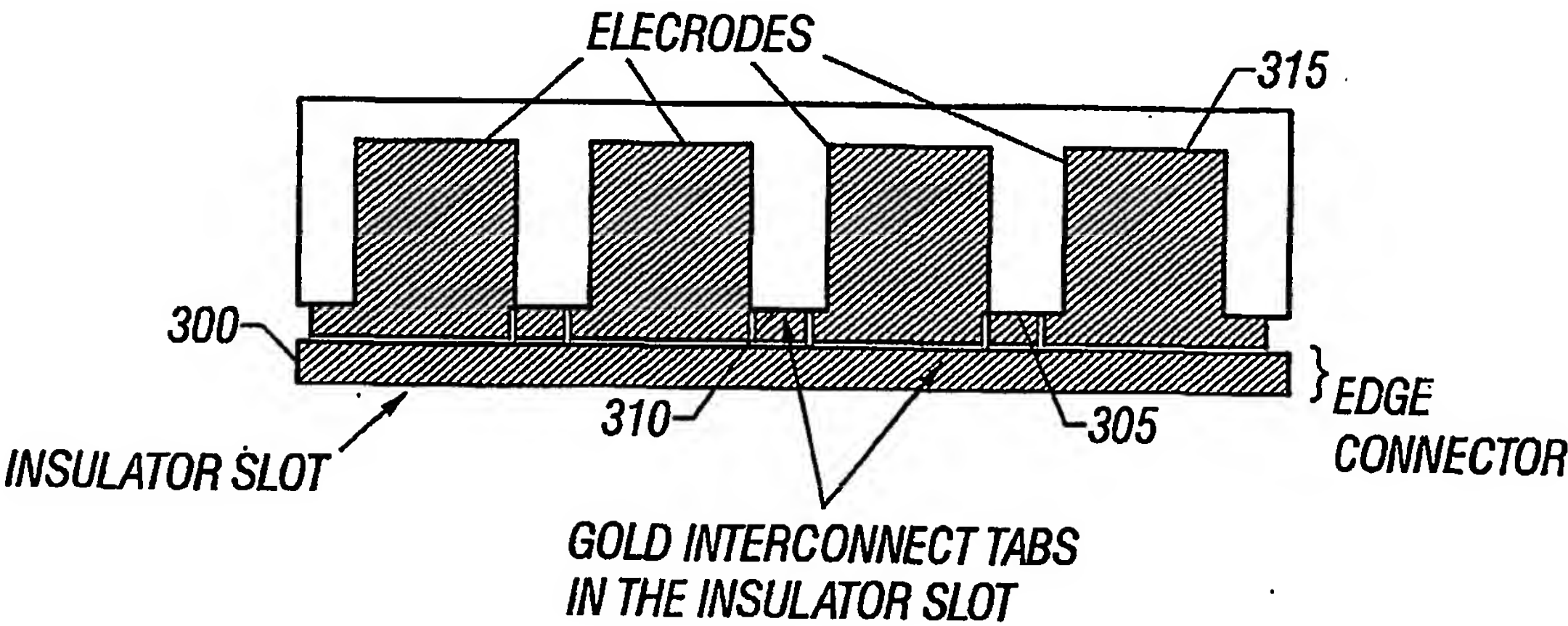


FIG. 3

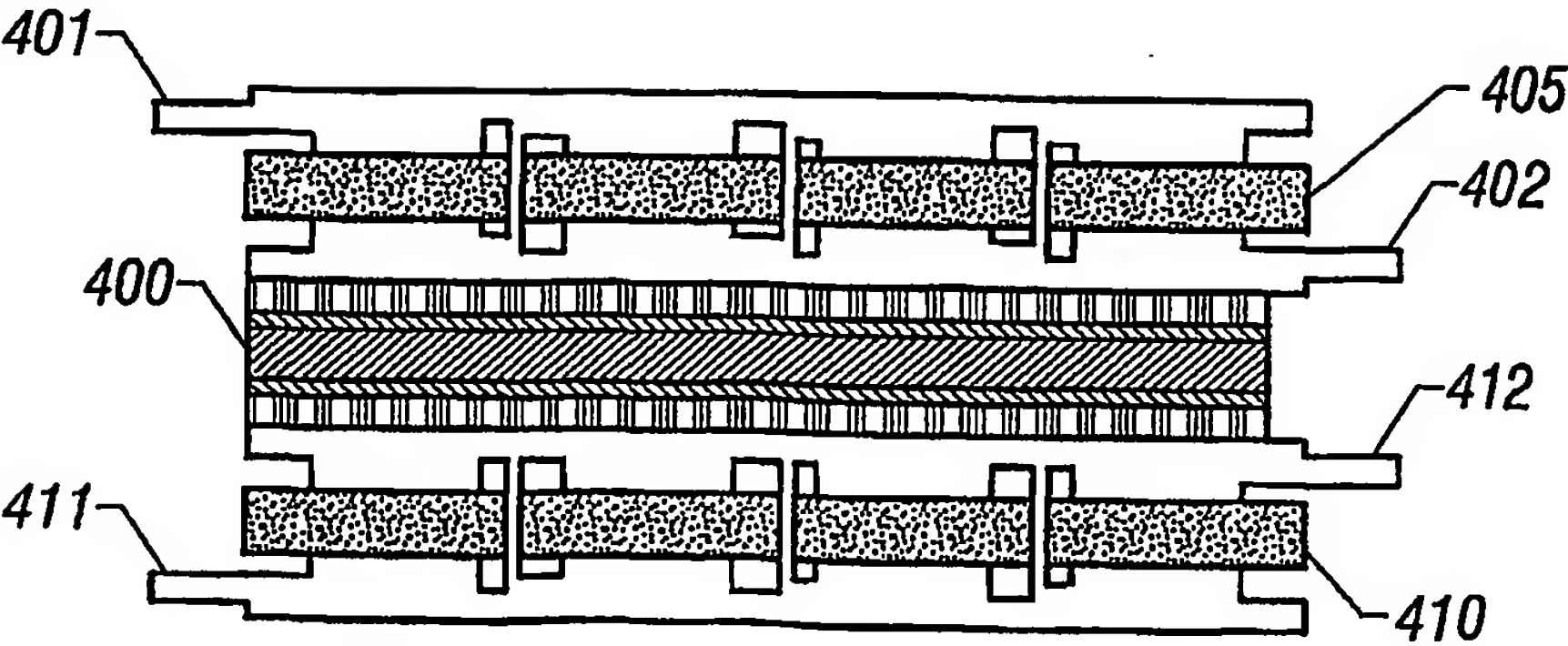


FIG. 4